



# Improving Well Water Quality Using Activated Carbon Derived from Ampel Bamboo Waste for Iron (Fe) and Manganese (Mn) Removal

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## ABSTRACT

The quality of well water used for consumption is highly important for community development. High concentrations of iron (Fe) and manganese (Mn) in well water can cause health problems. Therefore, in this study, we converted Ampel bamboo waste into activated carbon and used it to improve the well water quality in Bantul, Yogyakarta, Indonesia, with a focus on removing Fe and Mn, which typically exceeds the Indonesian water safety standards for daily use. Activated carbon from Ampel bamboo waste was produced at 900°C for 90 min. Fe and Mn removal from well water was conducted using a completely randomized design with two treatment factors: activated carbon dose (0.5, 1.25, and 2 g/L) and contact time (15, 45, and 75 min). Well water was obtained from Pelemsewu, Bantul Regency, Special Province, in Yogyakarta, Indonesia. The results showed that 2 g per liter of well water (g/L) of activated carbon and a contact time of 75 min were the most effective conditions for removing Fe and Mn, and reduced the concentrations of iron Fe and Mn by 94.72% (0.56–0.03 mg/L) and 49.72% (0.81–0.41 mg/L), respectively. After treatment, the water was suitable for drinking and met the Indonesian clean water quality standards outlined in the Minister of Health Regulation No. 2/2023.

**Keywords:** activated carbon, bamboo waste, iron (Fe), manganese (Mn), well water

## 1. INTRODUCTION

Water is a highly important resource for humans and other living organisms. It is used in households, for industrial and agricultural purposes, and for recreational activities such as fishing (Alkhatib *et al.*, 2015; Yazid *et al.*, 2021). Ensuring the quality of well water for consumption is essential for community development, as

water may be contaminated and must undergo proper treatment before being supplied to households as tap water (Yang *et al.*, 2019). High iron (Fe) concentrations can cause health problems related to mutations in genes responsible for absorbing iron and cause hemochromatosis, which can lead to liver, heart, and pancreas damage. Furthermore, it could be a precursor for diseases related to the heart and central nervous system,

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liver cirrhosis, diabetes, and nausea (Das *et al.*, 2023).

Drinking water quality guidelines and standards established by the Indonesian Government are designed to ensure clean, safe, and high quality water for human consumption and to protect public health. According to the Minister of Health Regulation of Indonesia No. 2/2023, clean water must meet the established physical, chemical, and biological quality parameters to ensure it is safe for human consumption (Ministry of Health, 2023). These water quality requirements are in accordance with the World Health Organization (WHO) guidelines for drinking-water quality (WHO, 2022).

In reality, in the Special Province of Yogyakarta, Indonesia, water quality varies, and has particularly high Fe and Mn concentrations, which is a critical problem; for example, in Pelemsewu, Bantul District, Special Province of Yogyakarta, the water has high levels of Fe and Mn and appears highly turbid, but it is still used for daily needs such as drinking, bathing, and washing. The high Fe concentration in the water leaves reddish-orange stains on bathroom fixtures, such as bathtubs, showers, and faucets, which are difficult to remove, and causes a decline in the color quality of washed clothes; high Fe concentrations cause clothes to turn red, yellow, or brown, making them appear dirty and unhygienic (Chaturvedi *et al.*, 2014; Das *et al.*, 2023). However, adsorption using activated carbon—an approach that has been used for decades and proven effective in removing heavy metals from water—can be applied to address this problem (Saeed *et al.*, 2020). Activated carbon adsorbs substances because of its high porosity, large surface area, and small particle size (Dungani *et al.*, 2022; Singh *et al.*, 2008; Sutapa *et al.*, 2024b). Activated carbon is an appropriate filter material for hazardous materials and water and can be used to improve the quality of drinking water and wastewater (Schröder *et al.*, 2007).

The production of bamboo chopsticks produces solid waste that cannot be utilized for the chopsticks. This waste is highly abundant and is generally not utilized optimally (Chen *et al.*, 2017). Studies have been con-

ducted on the conversion of bamboo waste into solid biofuel as an alternative to burning and co-firing with other resources (Chen *et al.*, 2017). However, bamboo waste can also be used as a raw material to produce activated carbon (Choy *et al.*, 2005; Mahanim *et al.*, 2011; Zhang *et al.*, 2014). Activated carbon is a suitable adsorbent for removing trace contaminants from air, soil, and water because of its high adsorption capacity (Skoczko and Guminski, 2024). The use of Ampel bamboo waste from the chopsticks industry to produce activated carbon is an innovative and practical approach, given the sustainability and biological availability of bamboo as a renewable resource; bamboo is a fast-growing plant (ranging from 3–5 years) with a high level of production, making it a major global biomass resource (Jeon *et al.*, 2018; Sumardi *et al.*, 2022). Furthermore, bamboo has been cultivated in Indonesia in forests and with edible crops in residential gardens (Park *et al.*, 2018).

In this study, charcoal was produced from Ampel bamboo waste as an eco-friendly material (Hwang and Oh, 2022). The charcoal was then converted to activated carbon and used to filter well water from the rural communities of Pelemsewu to improve its quality. The aim of this study was to determine the optimal dose and contact time of Ampel bamboo waste-derived activated carbon for improving well water quality. Specifically, we evaluated its effectiveness in removing Fe and Mn to ensure safe and high-quality water for human consumption.

## 2. MATERIALS and METHODS

Well water samples were obtained from Pelemsewu, Bantul District, Special Province, Yogyakarta, Indonesia. Ampel bamboo waste was obtained from the CV Jaya Abadi chopstick manufacturing factory in Tasikmalaya, West Java, Indonesia. This study was conducted using a completely randomized design with two treatment factors: activated carbon dose (0.5, 1.25, and 2 g/L) and

contact time (15, 45, and 75 min) with three replicates. Analysis of variance (ANOVA) was performed, followed by Tukey's honest significant difference test to test for significance analysis.

## 2.1. Preparation of bamboo activated carbon

### 2.1.1. Ampel bamboo waste carbonization

The bamboo waste was cut into smaller rectangular pieces ( $2 \times 7$  cm dimensions) and air-dried until the moisture content reached 10%-15%. The air-dried bamboo was carbonized at a temperature of  $400^{\circ}\text{C}$  for  $\pm 3$  h using an electric retort. The carbonized bamboo (charcoal) was ground into powder form and filtered through a 10-mesh sieve, retaining the particles on a 20-mesh sieve to ensure a homogeneous particle sizes (Fig. 1).

### 2.1.2. Carbon activation

Carbon Activation was conducted as follows: 52.61 g of bamboo charcoal was inserted in a 200 mL curable porcelain cup and heated at  $900^{\circ}\text{C}$  for 90 min in a furnace (FB 1410M-33; single setpoint with a capacity of 2.1 L; power consumption, 1,520 W; temperature

range,  $100^{\circ}\text{C}$ - $1,100^{\circ}\text{C}$ ; temperature stability,  $\pm 5.0$  at  $1,000^{\circ}\text{C}$ ; and electrical requirements of 240 V 50/60 Hz). This process was repeated for five replicates. The characteristics of the resulting activated carbon were evaluated and subsequently used in batch experiments to remove Fe and Mn from well water and improve the quality.

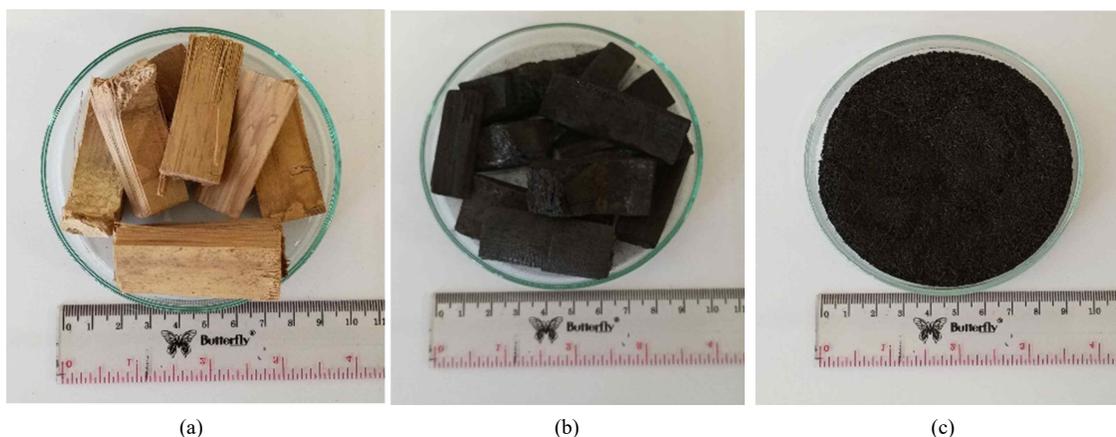
## 2.2. Characterization of bamboo activated carbon

### 2.2.1. Scanning electron microscope images observation

Morphological changes in the charcoal surface before (batch 1) and after (batch 2) activation were investigated using a Tabletop scanning electron microscope (SEM) TM4000Plus (Hitachi High-Technologies, Tokyo, Japan) at an acceleration voltage of 10 kV. This instrument allowed the analysis of samples without the need for conductive coating.

### 2.2.2. X-ray diffraction analysis

X-ray diffraction (XRD) analysis of charcoal and activated charcoal was performed using a Bruker D2 Phaser X-ray Diffractometer (Bruker, Berlin, Germany)



**Fig. 1.** Conversion of Ampel bamboo waste into charcoal and activated carbon. (a) Raw material, (b) bamboo charcoal, and (c) activated carbon.

operating at 30 kV and 10 mA, equipped with Cu K $\alpha$  radiation ( $\lambda = 1.54184 \text{ \AA}$ ). Diffraction patterns were collected over a  $2\theta$  range of  $5^\circ$ – $80^\circ$ .

### 2.3. Batch experiment to remove Fe and Mn

A total of 1,000 mL of well water was mixed with Ampel bamboo waste-derived activated carbon at doses of 0.9, 1.2, and 1.5 g/L, and contact times of 15, 45, and 75 min, with three replicates of each combination. The resulting solution was mixed and stirred in an Erlenmeyer flask using a magnetic stirrer (C\_MAG HS 7, IKA, Shanghai, China) at a temperature of  $30^\circ\text{C}$  and a speed of 1,000 rpm, and then filtered through filter paper, following the method described by Subroto *et al.* (2025), with slight modifications. This method used the following variables: dose ranging from 0.2–0.7 g/L, contact time ranging from 30–150 min, and stirring speed between 100 and 400 rpm.

The filtrate was then analyzed for quality parameters, including Fe, Mn, color, pH, turbidity, and water hardness. The treated well water was compared with the Indonesian drinking water quality standards according to the Minister of Health Regulation No. 2/2023.

### 2.4. Quality evaluation of treated well water

The parameters and standards used to evaluate well water quality after treatment with activated carbon include Fe and Mn concentrations (SNI 6989.04-2019), color (SNI 6989.80-2011; Badan Standardisasi Nasional, 2011), pH (SNI 6989.11-2019; Badan Standardisasi Nasional, 2019a), turbidity (SNI 06-6989.25-2005; Badan Standardisasi Nasional, 2005), and total hardness (SNI 06-6989.12-2004; Badan Standardisasi Nasional, 2004).

#### 2.4.1. Iron (Fe) and manganese (Mn) concentrations

Fe and Mn measurements were performed using an

atomic absorption spectrophotometer (SSA Hitachi Z-2000, 240 V Model 7 JO-8024) and following the potentiometric method (SNI 6989-04-2019; Badan Standardisasi Nasional, 2019b). The following equation was used to determine Fe and Mn concentrations:

$$\text{Total metal concentrations (Fe or Mn; mg/L)} = C \times Fp \quad (1)$$

where  $C$  is the metal concentrations (mg/L) obtained from measurement results and  $Fp$  is the dilution factor.

#### 2.4.2. Color content

Color content was evaluated using a spectrophotometer at wavelengths ranging from 450–465 nm using a standard Pt-Co solution to measure the true color of the test sample solution based on Beer's law.

The following equation was used to determine color content:

$$\text{Color (Pt-Co)} = D \times Fp \quad (2)$$

where  $D$  is the value obtained from the calibration curve (expressed in Pt-Co units) and  $Fp$  is the dilution factor.

#### 2.4.3. pH

pH was measured based on potentiometric hydrogen ion ( $\text{H}^+$ ) activity using a pH meter LUTRON PH-207. Briefly, the electrode was rinsed with mineral-free water and dried with soft tissue paper. The electrode was then submerged in the test sample until the pH meter provided a correct and stable reading. The readings were then recorded. Temperature was measured simultaneously with the pH measurement, and the results were recorded in a laboratory worksheet. Finally, the electrode was rinsed with mineral-free water again.

#### 2.4.4. Turbidity

A nephelometer was used to measure water turbidity. The amount of light that was absorbed and refracted in

the test sample was compared with that of a standard (clear) suspension.

The following equation was used to calculate turbidity:

$$\text{Turbidity (NTU)} = T \times Fp \quad (3)$$

where  $T$  is the turbidity (NTU) of the diluted sample and  $Fp$  is the dilution factor.

#### 2.4.5. Total hardness

The total hardness of the water and wastewater was determined using the EDTA titrimetric method, with a lower limit of 5 mg/L. This method was used for the colorless water samples.

The following equation was used to test water hardness:

$$\text{Total hardness (mg CaCO}_3\text{/L)} = 1,000/V_{C.u} \times V_{EDTA(a)} \times M_{EDTA} \times 100 \quad (4)$$

where  $V_{C.u}$  is the volume of test sample solution (mL),  $V_{EDTA(a)}$  is the average volume of Na<sub>2</sub>EDTA standard solution for total hardness titration (mL), and  $M_{EDTA}$  is

the molarity of Na<sub>2</sub>EDTA standard solution for titration (mmol/mL).

## 3. RESULTS and DISCUSSION

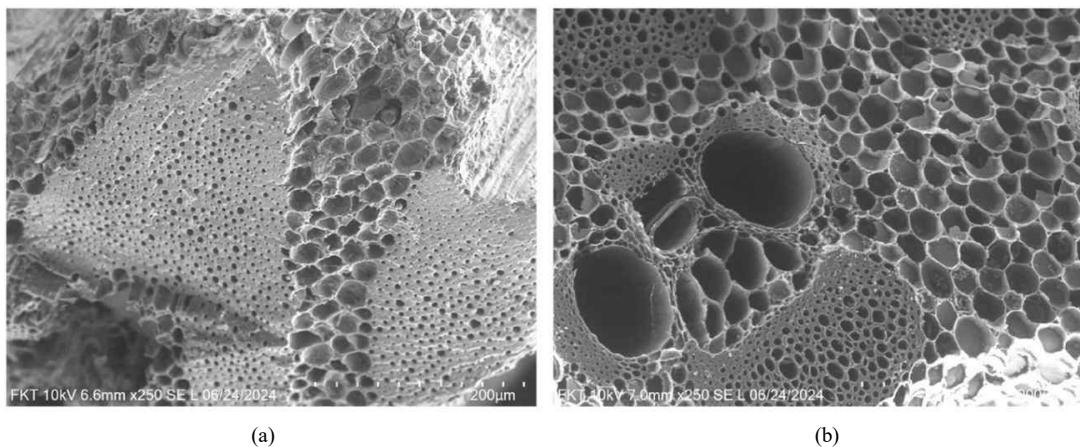
### 3.1. Morphological characterization

Fig. 2 shows a comparison of the surface morphologies of the charcoal and activated carbon. The SEM images show that activated carbon exhibits significantly higher porosity than ordinary charcoal. This enhanced porosity and removal of the inert material in the cell lumens of the activated carbon [Fig. 2(b)] indicate its large surface area, which is essential for its effectiveness in adsorption applications (Zhu *et al.*, 2024).

Analysis of the SEM cross-sectional images showed that, after activation, the sclerenchyma cells and phloem were cleared of inert material, resulting in open pores and clearly defined cell walls (Fig. 2).

### 3.2. X-ray diffraction analysis

XRD analysis was conducted to investigate the crystalline structures of the bamboo charcoal and acti-



**Fig. 2.** Scanning electron microscopic (SEM) cross-sectional images of charcoal at 250 × magnification (a) before and (b) after activation treatment.

vated carbon (Lazzarini, 2017). The patterns generally revealed an amorphous structure for charcoal and activated carbon. However, the distinct peaks indicate the presence of crystalline characteristics (Fig. 3). Both materials exhibit distinct peaks at approximately  $2\theta = 23^\circ$ , corresponding to a graphite-like microcrystalline structure (002) lattice plane. In contrast to charcoal, activated carbon displayed sharper peaks, indicating a greater regularity in its crystalline structure due to the higher-temperature treatment. Additionally, activated carbon shows a peak at approximately  $2\theta = 43^\circ$ , which corresponds to the (100) lattice planes of a graphite-like microcrystalline structure (Wang *et al.*, 2014). To further quantify the structural changes, the crystallinity index (CrI) was measured using the method described by Segal *et al.* (1959):  $CrI = [(I_{002} - I_{am}) / I_{002}] \times 100\%$ . The analysis revealed that CrI increased from 40.49% in bamboo charcoal to 51.06% in the activated carbon.

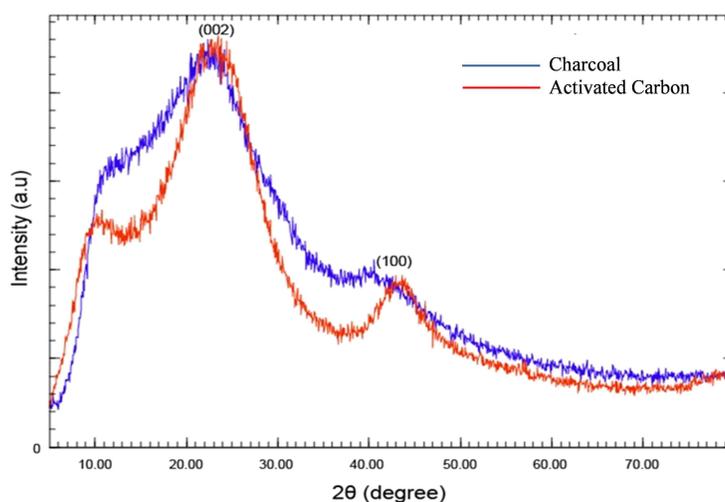
Based on the peak intensity, this XRD pattern was highly similar to the XRD pattern of the activated carbon of palm kernel shells collected from the palm oil industry in Muara Enim (Damayanti *et al.*, 2025).

Furthermore, the XRD pattern of activated carbon from bamboo waste was similar to that of bamboo carbon activated using steam-based methods (Zhu *et al.*, 2024). These findings imply that before heat activation, the charcoal exhibited an amorphous structure, and subsequent thermal treatment significantly improved its graphitic characteristics (Wang *et al.*, 2018).

Finally, the interlayer spacing ( $d_{002}$ ) was checked using Bragg's equation [ $d_{002} = \lambda / (2 \sin \theta)$ ]; Almgren *et al.*, 2022). The analysis shows that the spacing decreased from 0.3897 nm for charcoal to 0.3798 nm for activated carbon. This indicates that the activation process allowed the graphite sheet to pack more tightly.

### 3.3. Proximate Ampel bamboo waste-derived activated carbon

This study used the most effective method to produce activated carbon from Ampel bamboo waste, which was described by Sutapa and Ramba (2021). The activated carbon originated from the activation of Ampel bamboo waste charcoal at a temperature of  $900^\circ\text{C}$  for 90 min.



**Fig. 3.** X-ray diffraction (XRD) pattern of charcoal and activated carbon. Labels on the peaks indicate the lattice planes of the graphite-like microcrystalline structure.

Table 1 shows the quality parameters of the activated carbon from Ampel bamboo waste, that is, moisture content, volatile matter, and fixed carbon content, compared with the Indonesian quality standard SNI 06-3730-1995.

Moisture content refers to the amount of water physically bound to the activated carbon under normal conditions. In contrast, carbon ash refers to the residue that remains after the material is burned. Ash contains inorganic material that is bound to the activated carbon, and generally ranges from 2%–10%. The lower the ash content, the better the activated carbon, as high ash content can lead to increased hydrophilicity and catalytic effects, resulting in a restructuring process during the regeneration of used activated carbon (Nurul'ain, 2007). High levels of volatile matter indicate that the surface of the activated carbon contains non-carbon compounds and substances, which originate from the interaction of carbon with water vapor; therefore, a low volatile matter content implies high porosity of the adsorbent because less volatile matter remains clogged in the pores (Haji et al., 2013; Qian et al., 2007). This lower volatile matter content is caused by the removal of volatile compounds during carbonization at high temperatures

(Ju et al., 2020). The fixed activated carbon content in this study was 72.41%, which complied with the SNI 06-3730-1995 standard. A higher value of fixed carbon leads to better iodine absorption. These results align with those of a study conducted by Sutapa et al. (2024a) on activated carbon from the sapwood waste of fast-growing teak. They reported that the highest iodine adsorption value was found at high fixed-carbon values.

### 3.4. Ampel bamboo waste activated carbon adsorption capacity

The characteristics of the adsorption capacity of Ampel bamboo waste-activated carbon include the adsorption capacities for benzene (%), iodine (mg/g), and methylene blue (mg/g). Its adsorption capacity for benzene was 8.32%. The capacity for benzene adsorption is related to the surface area of activated carbon; greater benzene absorption indicates a higher level of porosity in the activated carbon (Isinkaralar, 2022). The adsorption capacity for benzene is an indicator of the quality of activated carbon in absorbing pollutant gases (Mohammad-Khan and Ansari, 2009), thus, the adsorption capacity for benzene is not a relevant indicator for

**Table 1.** Quality characteristics of Ampel bamboo waste-derived activated carbon compared with SNI Standard 06-3730-1995 requirements

| Quality parameters*                          | Result number | SNI Standard 06-3730-1995 | Explanation   |
|----------------------------------------------|---------------|---------------------------|---------------|
| Yield (%)                                    | 73.89         | -                         | -             |
| Moisture content (%)                         | 6.344         | ≤ 15                      | Fulfilled     |
| Volatile matter (%)                          | 15.49         | ≤ 25                      | Fulfilled     |
| Ash content (%)                              | 12.10         | ≤ 10                      | Not fulfilled |
| Fixed carbon content (%)                     | 72.41         | ≥ 65                      | Fulfilled     |
| Adsorption capacity of benzene (%)           | 8.32          | ≥ 25                      | Not fulfilled |
| Adsorption capacity of iodine (mg/g)         | 752.82        | ≥ 750                     | Fulfilled     |
| Adsorption capacity of methylene blue (mg/g) | 97.90         | ≥ 120                     | Not fulfilled |

\* Data from Sutapa and Ramba (2021).

the removal of Fe and Mn.

The iodine adsorption capacity of the activated carbon was 752,818 mg/g, meeting the SNI 06-3730-1995 quality standard. According to Idrus *et al.* (2023), activated carbon with a high iodine adsorption capacity that meets quality standards can absorb adsorbates or dissolved substances. Based on the iodine adsorption capacity, the surface area of the activated carbon was calculated according to an equation described by Mianowski *et al.* (2007):  $SA = 0.986 \times IN$ , where  $SA$  is the surface area and  $IN$  is the iodine absorption number. The calculated surface area of activated carbon was 742 m<sup>2</sup>/g. Thus, the high iodine adsorption capacity of activated carbon indicates a high capability to absorb Fe and Mn. Therefore, activated carbon from Ampel bamboo waste can be used as an alternative water filter media to improve water quality. As shown in Table 1, the methylene blue adsorption capacity was 97.90 mg/g, which does not meet the SNI 06-3730-1995 quality standard. The magnitude of methylene blue absorption

capacity is indicative of the ability of activated carbon to absorb colored liquid material solutions with a molecular size of 15 Å or 1.5 nm (Alimah, 2017). Although the activated carbon did not meet the absorption standard with methylene blue, the produced Ampel bamboo-activated carbon removed some color from the well water.

### 3.5. Impact of activated carbon treatment on water quality

Tables 2 and 3 show the results of using activated carbon from Ampel bamboo waste to improve well water quality. After treatment, Fe and Mn concentrations were reduced. Furthermore, quality improvements were observed in terms of pH, color, turbidity, and hardness. This result showed the effectiveness of activated carbon as an absorbent for improving water quality. Fig. 4 shows the effect of the Ampel bamboo waste-activated carbon on the color of the Pelemsewu well water before

**Table 2.** The quality of Pelemsewu well water based on various *in situ* parameters

| Adsorbent doses (g/L) | Contact time (min)                                                    | Iron (Fe; mg/L) | Manganese (Mn; mg/L) | Color (TCU) | pH  | Turbidity (NTU) | Calcium hardness (mg/L) |
|-----------------------|-----------------------------------------------------------------------|-----------------|----------------------|-------------|-----|-----------------|-------------------------|
| 0 (before treatment)  | 0                                                                     | 0.56            | 0.81                 | 20          | 7.5 | 100             | 202.99                  |
|                       | 15                                                                    | 0.08            | 0.63                 | 1           | 8.2 | 0.53            | 185.74                  |
|                       | 0.5                                                                   | 45              | 0.03                 | 0.61        | 3   | 8.3             | 183.08                  |
| 0.5                   | 75                                                                    | 0.04            | 0.66                 | 3           | 8.5 | 0.63            | 196.35                  |
|                       | 15                                                                    | 0.03            | 0.65                 | 3           | 8.4 | 0.60            | 197.02                  |
|                       | 1.25                                                                  | 45              | 0.04                 | 0.61        | 1   | 8.4             | 187.06                  |
| 1.25                  | 75                                                                    | 0.08            | 0.56                 | 2           | 8.6 | 1.00            | 169.15                  |
|                       | 15                                                                    | 0.09            | 0.51                 | 1           | 8.7 | 1.13            | 188.39                  |
|                       | 2                                                                     | 45              | 0.06                 | 0.50        | 1   | 8.8             | 185.07                  |
| 2                     | 75                                                                    | 0.07            | 0.41                 | 1           | 8.9 | 1.10            | 177.11                  |
|                       | Average                                                               | 0.06            | 0.57                 | 1.78        | 8.5 | 0.79            | 185.44                  |
|                       | Regulation Standard of the Minister of Health of Indonesia No. 2/2023 |                 | 0.2                  | 0.1         | 10  | 6.5-8.5         | < 3                     |

**Table 3.** Analysis of variance results for adsorbent doses and contact times on the quality parameters of well water

| Source              | DF | Iron (Fe) | Manganese (Mn) | Color | pH | Turbidity | Total hardness |
|---------------------|----|-----------|----------------|-------|----|-----------|----------------|
| Adsorbent doses (A) | 2  | NS        | **             | NS    | ** | **        | NS             |
| Contact time (C)    | 2  | NS        | **             | NS    | ** | NS        | NS             |
| A × C               | 4  | NS        | **             | NS    | NS | NS        | NS             |
| Error               | 18 |           |                |       |    |           |                |
| Total               | 27 |           |                |       |    |           |                |

NS: non-significant ( $p > 0.01$ ); \*\*  $p < 0.01$ .

DF: degree of freedom.



**Fig. 4.** Well water from Pelemsewu before and after treatment with bamboo Ampel waste-derived activated carbon.

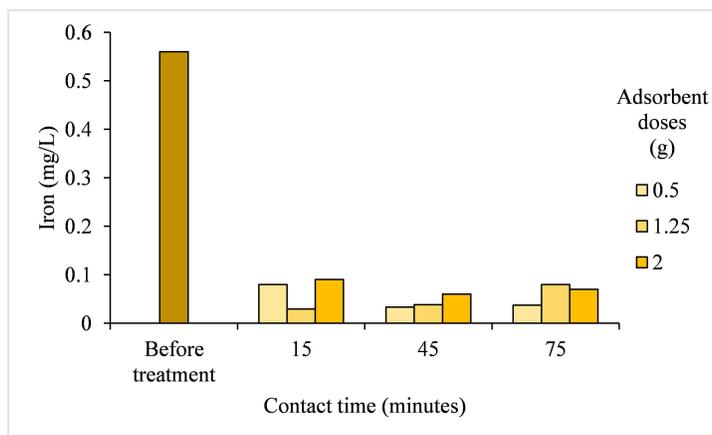
and after treatment. The well water, which was initially cloudy, became clear after treatment.

### 3.6. Impact of activated carbon treatment on the Fe concentration in well water

High Fe concentrations in groundwater can cause health problems in the skin and the digestive, respiratory, and nervous systems (Achary, 2014). High Fe concentrations in well water can affect its quality; for example, the water has a metallic taste, unpleasant odor, and discoloration (Subroto *et al.*, 2025). Kumar and Sinha (2018) reported that in Moradabad City, India, the

highest Fe concentrations in well water were recorded during the summer months, and slightly lower concentrations were recorded in winter. Water containing high Fe levels causes stains on clothing and sanitary equipment and gives a bitter astringent taste.

The Fe concentration in the Pelemsewu well water was 0.56 mg/L. After treatment with activated carbon, the average Fe concentration in the well water was 0.06 mg/L, ranging from 0.03–0.09 mg/L (Table 2). Based on the ANOVA, the Fe concentration in the well water after treatment was influenced by the interaction between the adsorbent dose and contact time; however, they showed no significant effect, but Fe levels in the well water were still reduced. Fig. 5 shows that treatment with an activated carbon adsorbent dose of 0.5 g with a contact time of 45 min decreased Fe concentration by 94.73% (from 0.56–0.03 mg/L). The Fe concentration in the well water after treatment with Ampel bamboo waste-activated carbon showed a very high reduction because of the small distance between Fe ions, making it easier for them to penetrate the pores of the activated carbon (bin Jusoh *et al.*, 2005). The results showed that well water treatment using activated carbon from Ampel bamboo waste produced the lowest reduction in Fe concentration (80.33%). This result was affected by the higher iodine absorption capacity (752.82 mg/g). Furthermore, according to Dastgheib and Rockstraw (2002), activated carbon



**Fig. 5.** Iron (Fe) concentration (mg/L) in well water after treatment with Ampel bamboo waste-derived activated carbon.

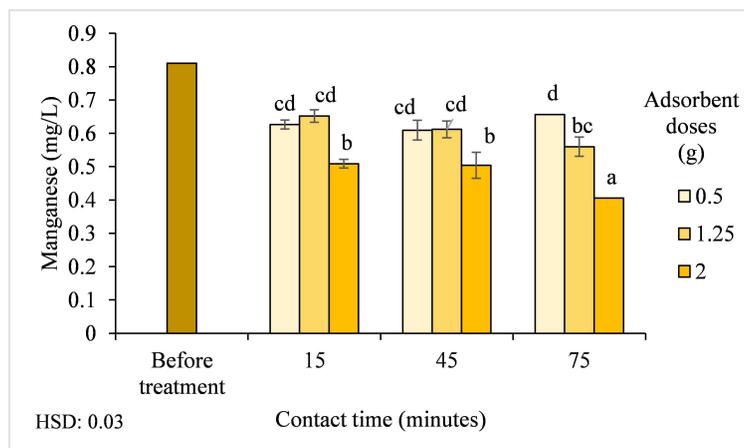
has a higher adsorption capacity for Fe because of its electronegativity, which is similar to its ability to attract electrons; higher electronegativity corresponds to a higher adsorption capability for metal ions. In this case, Fe, which has a positive charge, is attracted to a negative charge on the activated carbon surface.

This value is slightly lower than that of coal-based activated carbon modified with  $\text{NiFe}_2\text{O}_4$ , which can reduce Fe concentration by 96.12% (Subroto *et al.*, 2025). The results showed that well water treated with Ampel bamboo waste-derived activated carbon at doses of 0.5, 1.25, and 2 g/L and contact times of 15, 45, and 75 min met the Indonesian standards of the Minister of Health Regulation No. 2/2023 for Fe, with a maximum allowable concentration of 0.2 mg/L. These findings show Ampel bamboo waste-derived activated carbon is effective in improving the quality of water from the Pelemsewu wells. Therefore, treatment using activated carbon is a viable solution for improving the quality of water containing high Fe concentrations.

### 3.7. Impact of activated carbon treatment on the Mn concentration in well water

The ability of activated carbon to remove Mn from

well water is an important characteristic because Mn pollution pose significant risks to public health. Mn at concentrations  $> 300 \mu\text{g/L}$  have been shown to have detrimental neurological effects on infants (Dvorak and Schuerman, 2021). The initial Mn concentration of the Pelemsewu well water was 0.81 mg/L, but after treatment, the average Mn concentration was reduced to 0.57 mg/L, ranging from 0.41–0.66 mg/L (Table 2). Based on the ANOVA results, the Mn concentration in the treated well water was significantly affected by the interaction between the adsorbent dose and contact time ( $p < 0.01$ ). Fig. 6 shows that Mn concentration decreased by 49.72% with increasing adsorbent doses and contact times. The largest decrease occurred after treatment with 2 g/L activated carbon for 75 min. Badawi *et al.* (2017) stated that as the dose increases, the adsorption on the adsorbent surface area increases, thereby reducing the amount of pollutants. This result is similar to that of Subroto *et al.* (2025), who stated that both activated carbon and activated carbon/ $\text{NiFe}_2\text{O}_4$  composites have similar affinities for reducing Mn concentrations. The mixing of activated carbon and  $\text{NiFe}_2\text{O}_4$  improved the separation process and increased the adsorption capacity. Ampel bamboo waste-derived activated carbon has a porous structure that increases its absorption capacity for



**Fig. 6.** Manganese (Mn) concentration (mg/L) in well water after treatment with Ampel bamboo waste-derived activated carbon. <sup>a-d</sup> Groups in a homogeneous subset. HSD: honestly significant difference.

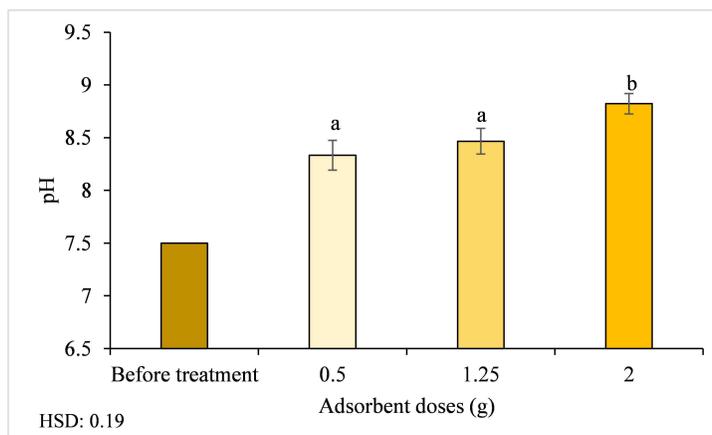
Mn ions (Budinova *et al.*, 2009). This high absorption capacity for Mn is attributed to the positively charged Mn ions, which interact and bind to functional groups on the surface of the activated carbon through chemical bonding (bin Jusoh *et al.*, 2005). Functional groups attract cations, promoting ion exchange. Although the results showed a reduction Mn concentration of Pelemsewu well water after treatment (0.81–0.57 mg/L), this concentration does not meet the Indonesian clean water quality standards of the Minister of Health Regulation of Indonesia No. 2/2023 for Mn, which is 0.1 mg/L.

Another factor contributing to the different activated carbon adsorption capacities for Mn and Fe ions is their ionic radii. Fe has a smaller ionic radius than Mn; the smaller ionic radius of Fe allows them to easily penetrate the micropores of activated carbon.

### 3.8. Impact of activated carbon treatment on pH of well water

The pH can be used as an indicator of water hardness. Pure water has a pH of 7, surface water has a pH range of 6.5–8.5, and groundwater has a pH range of 6–

8.5. The biological effects of low pH include to gill damage, mucus buildup on the gills, stunted growth, problems with ion regulation, and reproductive failure, which leads to a reduction in the number of aquatic species in the environment and replacement of acid-sensitive species with acid-tolerant species (Dirisu *et al.*, 2016; WHO, 2022). The pH of water influences the rate of metal corrosion and the efficiency of disinfection. Thus, the pH of water entering the distribution system for households or industries must be maintained to minimize the corrosion of water channels and pipes. Corrosion and failure of water pipes can contaminate drinking water and affect its taste, odor, and appearance (WHO, 2022). Furthermore, the pH is an important factor influencing the adsorption of Fe (II) ions (Karim *et al.*, 2024). The initial pH of the Pelemsewu well water was 7.5, and after treatment with activated carbon from Ampel bamboo waste the average pH increased to 8.5, ranging from 8.2–8.9 (Table 2). Based on the ANOVA results, the adsorbent doses significantly affected the pH ( $p < 0.01$ ). Fig. 7 shows that the higher the dose of activated carbon, the higher the pH value of the water. The activated carbon influenced the pH by decomposing the metal elements in the well water into



**Fig. 7.** The pH of well water after treatment with Ampel bamboo waste-derived activated carbon. <sup>a,b</sup> Groups in a homogeneous subset. HSD: honestly significant difference.

metal and hydroxide ions (OH<sup>-</sup>). Metal ions are attracted to the surface of activated carbon, which interacts and binds with acid groups so that H<sup>+</sup> ions are reduced, and only OH<sup>-</sup> ions remain (Heriyani and Mugisidi, 2016). After treatment with Ampel bamboo waste activated carbon, the pH of Pelemsewu well water met the Indonesian drinking water quality standards (range of 6.5–8.5) of the Minister of Health Regulation No. 2/2023.

### 3.9. Impact of activated carbon treatment on color (TCU) of well water

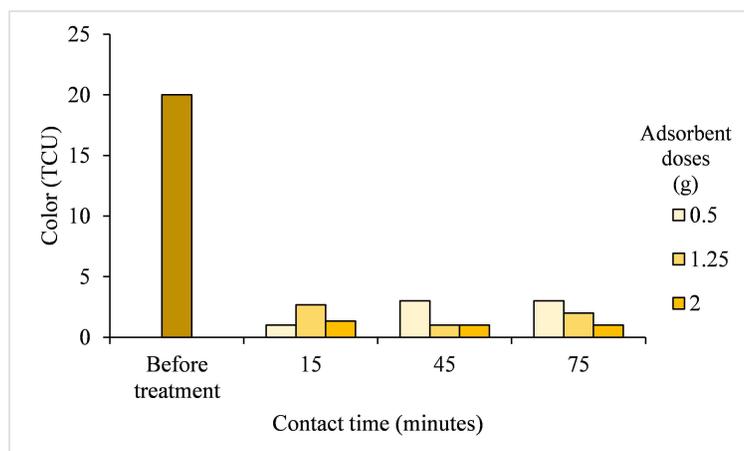
The color of water and wastewater is an important parameter of water quality as it affects photosynthesis by phototrophic organisms and influences dissolved oxygen content; thus, color changes can result in an ecological imbalance. In addition, some colored substances in water can be categorized as toxic or carcinogenic (Bao *et al.*, 2017). The color content of the well water of Pelemsewu Bantul was initially 20 TCU. After treatment with Ampel bamboo waste activated carbon, the average color content of the well water was reduced to 1.78 TCU, ranging from 1–3 TCU (Table 2). The color content of the well water after treatment was influ-

enced by the interaction between the activated carbon dose and contact time. The highest reduction in color content reached 95% (from 20–1 TCU), which occurred at adsorbent doses of 0.5 g and 1.25 g with a contact time of 15 min and at an adsorbent dose of 2 g/L with contact times of 15, 45, and 75 min (Fig. 8).

This considerable decrease in color content is correlated with a reduction in Fe concentration in the Pelemsewu well water after treatment because of the microporous structure of activated carbon, which absorbs chemical particles (Gürses *et al.*, 2006). The color content of the well water after treatment met the Indonesian quality standards of the Minister of Health Regulation of No. 2/2023, which is 10 TCU. This suggests that Ampel bamboo waste-derived activated carbon is a promising water purification medium.

### 3.10. Impact of activated carbon treatment on the turbidity (NTU) of well water

The higher the number of suspended solids in the water, the more turbid the water. Turbidity is a parameter used worldwide to describe the quality of drinking water (WHO, 2022). In this study, the initial turbidity of well water from Pelemsewu was 100 NTU. After treat-



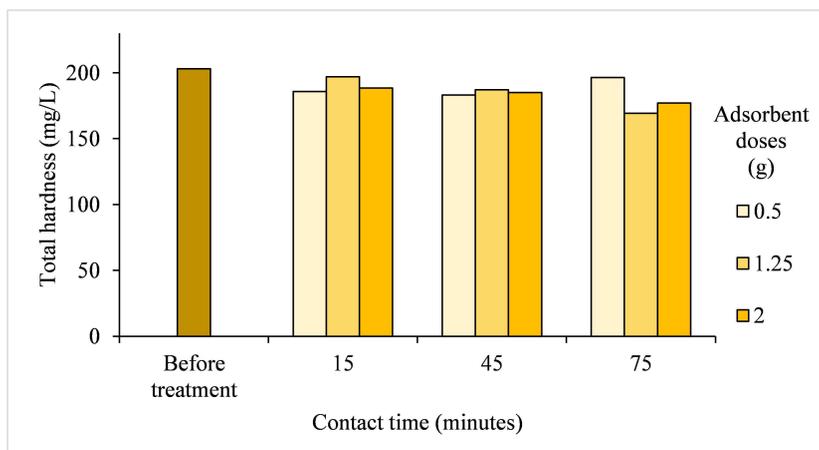
**Fig. 8.** Color (TCU) of well water after treatment with Ampel bamboo waste-derived activated carbon.

ment with activated carbon from Ampel bamboo waste, the turbidity was reduced to 0.79 NTU, ranging from 0.43–1.13 NTU (Table 2), which is a significant reduction (> 99%). The reduction in turbidity was 99.47%, 99.26%, and 98.91% at adsorbent doses of 0.5, 1.25, and 2 g/L, respectively. This result is consistent with that of Subroto *et al.* (2025), who proved that coal-based activated carbon modified with NiFe<sub>2</sub>O<sub>4</sub> can reduce the turbidity of water. The reduction in turbidity observed in this study is greater than that reported by Lin *et al.* (2017), who found a 22.59% decrease using multi-layer water filtration at a flow velocity of ± 5 mL/min through Moso bamboo-derived activated carbon. This is likely due to the higher iodine absorption capacity of Ampel bamboo waste-derived activated carbon, which has a well-developed microporous structure that enhances its ability to remove suspended substances from the water (Budiman *et al.*, 2019).

The turbidity of well water treated with Ampel bamboo waste-derived activated carbon at doses of 0.5, 1.25, and 2 g/L and contact times of 15, 45, and 75 min, respectively, met the Indonesian quality standards of the Minister of Health Regulation No. 2/2023, which is a maximum turbidity of < 3 NTU.

### 3.11. Impact of activated carbon treatment on total hardness of well water

Hard water is harmful to human health and can cause diseases. Over the last five decades, the number of deaths related to cardiovascular disease due to the hardness of drinking water has increased (Akram and Rehman, 2018; Sengupta, 2013). The initial value of the total hardness of the Pelemsewu well water (202.99 mg/L) was significantly reduced after treatment with Ampel bamboo waste-derived activated carbon. The results showed that the average total hardness concentration of Pelemsewu well water after treatment was reduced to 185.44 mg/L, ranging from 165.15–197.02 mg/L (Table 2). The highest reduction in total hardness concentration (16.67%) occurred when the water was treated with an adsorbent dose of 1.25 g for 75 min (Fig. 9). This result is lower than that reported by Lin *et al.* (2017), who used Moso bamboo activated carbon in a multi-layer filtration system. In their study, activated carbon was placed in a glass funnel, and source water flowed from top to bottom under gravity at a controlled rate of  $5 \pm 2$  mL/min, resulting in a 50% reduction in total hardness in water from Chiayi Lantan Lake.



**Fig. 9.** Total hardness concentration (mg/L) in well water after treatment with Ampel bamboo waste-derived activated carbon.

The capacity of activated carbon to reduce total hardness is attributed to its porous surface; the higher the pore capacity or surface area, the greater the ability to reduce the total hardness in water (Adewuyi and Olabanji, 2022; Lin *et al.*, 2017; Nurhayati *et al.*, 2021).

#### 4. CONCLUSIONS

Based on our findings, we concluded that the interaction between the adsorbent dose and contact time of activated carbon and well water significantly affects the Mn concentration. The adsorbent dose also significantly influences the pH and turbidity of well water. The optimal treatment parameters for the Pelemsewu well water with Ampel bamboo waste-derived activated carbon was a 2 g/L adsorbent dose and 75-min contact time, which reduced Fe concentration by 87.50% (from 0.56–0.07 mg/L), Mn concentration by 49.91% (from 0.81–0.41 mg/L), color content by 95% (from 20–1 TCU), turbidity by 98.90% (from 100–1.10 NTU), and total hardness concentration by 12.75% (from 202.99–177.11 mg/L). However, the treatment only increased pH slightly by 18.67% (from 7.5–8.9). These results suggest that Ampel bamboo waste-derived activated

carbon is a promising material for improving well water quality, especially in the Bantul District, Special Province of Yogyakarta, Indonesia. After treatment, the Fe concentration in the well water met the Indonesian standards for safe drinking water outlined by the Ministry of Health Regulation No. 2/2023. In conclusion, activated carbon derived from Ampel Bamboo waste is an environmentally friendly and biodegradable material that can enhance the quality of well water for community use. For further development, future research should focus on designing a simple and practical water filtration system based on this material.

#### CONFLICT of INTEREST

No potential conflict of interest relevant to this article was reported.

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